

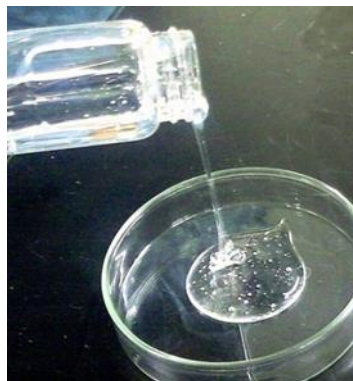
# Introduction of PEL series

## - Ion conductive additives for polymers

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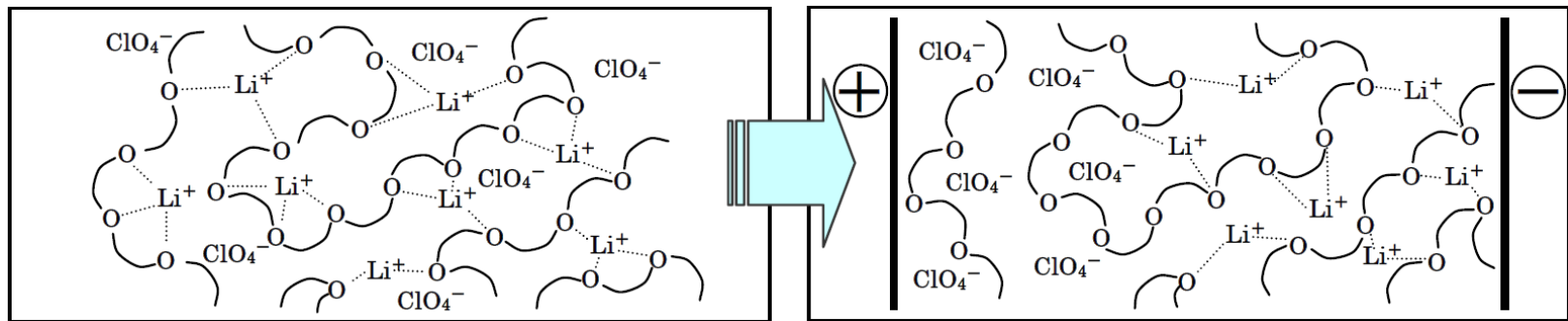
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1. A special class of complex between  $\text{Li}^+$  and poly(alkylene glycol)
2. Ion conductive w/ less humidity dependence
3. Transparent viscous liquid that can be added to polymer
  - several types are available for optimal miscibility w/ polymers
4. Applicable to anti-static packaging for electronics parts, rubber parts etc.



## 2. Conduction mechanism

1.  $\text{Li}^+$  from Li salts, *e.g.*  $\text{LiClO}_4$ , solvates with (coordinated by) oxygen atoms in poly(alkylene glycol) then dissociate w/  $\text{ClO}_4^-$ .
2.  $\text{ClO}_4^-$  is now easily mobile w/ assist of lattice vibration of the polymer chain while  $\text{Li}^+$  is fixed to the polymer.
3. When electric field is applied,  $\text{ClO}_4^-$  shows high mobility.
  - No assist of humidity is necessary.
  - shows high conductivity under low humidity environment
4. Form conductive pathways in non-conductive polymer such as PVC, poly(urethane), *etc.* when composite of PEL w/ them are made



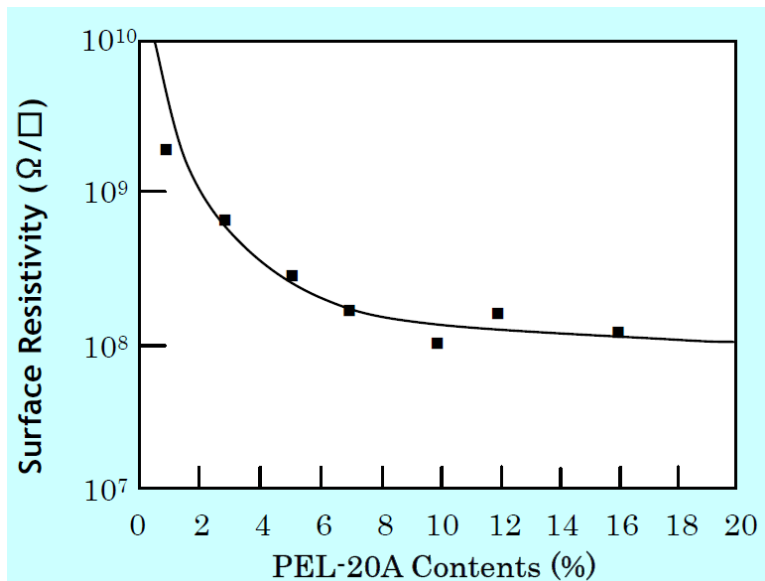


Fig. 1 Surface resistivity dependence on PEL 20A contents in poly(urethane)

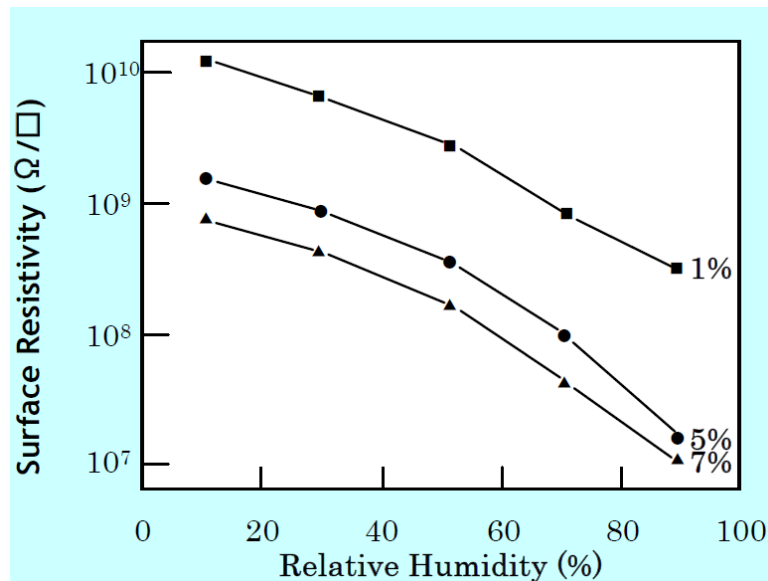


Fig. 2 Surface resistivity dependence of PEL 20A/poly(urethane) composite on relative humidity

PEL contents: 1, 5 and 7wt%

## 4. Lineup of PELs

|                                        | PEL-20A                             | PEL-100                                   | PEL-25                                                 | PEL-46                                 |
|----------------------------------------|-------------------------------------|-------------------------------------------|--------------------------------------------------------|----------------------------------------|
| appearance                             | transparent ~ pale yellowish liquid | transparent ~ pale yellowish liquid       | transparent ~ pale yellowish liquid                    | pale yellowish liquid                  |
| salt                                   | LiClO <sub>4</sub>                  | LiClO <sub>4</sub>                        | LiClO <sub>4</sub> /CF <sub>3</sub> SO <sub>3</sub> Li | organic boron complex Li salt          |
| salt content                           | 10wt%                               | 10wt%                                     | 15wt%                                                  | 30wt%                                  |
| matrix polymer                         | PEO-PPO copolymer                   | PPO                                       | PEO-PPO copolymer                                      | PEO-PPO copolymer                      |
| <i>M<sub>w</sub></i> of matrix polymer | ~1300                               | ~2000                                     | ~1300                                                  | ~1300                                  |
| hydroxyl number                        | ~78                                 | ~57                                       | ~74                                                    | ~58                                    |
| characteristics                        | standard grade                      | better miscibility w/ hydrophobic polymer | more conductive than 20A                               | Halogen free relatively low conductive |

\* By making use of hydroxyl end groups in the matrix polymer, it is possible to fix the matrix polymer into counterpart of composite polymer *e.g.* poly(urethane) with a crosslinking agent such as isocyanate.

| Polymer or resin                          | Miscibility w/ PEL |
|-------------------------------------------|--------------------|
| Poly(urethane), acrylic resin             | ○                  |
| Poly(ester), epoxy resin, PVC             | △                  |
| PE, PP, PS and other hydro carbon polymer | △ ~ ×              |
| PVDF                                      | ×                  |
| Silicone resin, styrene-acrylic resin     | ×                  |

### Legend

○: miscible

△: partially miscible

×: not miscible

## 6. Miscibility of PELs w/ various solvents

| solvent             | PEL-20A | PEL-100 | PEL-25 | PEL-46 |
|---------------------|---------|---------|--------|--------|
| water               | ○       | ×       | ○      | ○      |
| MeOH                | ○       | ○       | ○      | ○      |
| EtOH                | ○       | ○       | ○      | ○      |
| IPA                 | ○       | ○       | ○      | ○      |
| acetone             | ○       | ○       | ○      | ○      |
| MEK                 | ○       | ○       | ○      | ○      |
| NMP                 | ○       | ×       | ○      | ×      |
| acetonitrile        | ○       | ○       | ○      | ○      |
| EtOAc               | ○       | ○       | ○      | ○      |
| BuOAc               | ○       | ○       | ○      | ○      |
| propylene carbonate | ○       | ○       | ○      | ○      |
| <i>m</i> -xylene    | ○       | ○       | ×      | ○      |
| toluene             | ○       | ○       | ○      | ○      |
| hexane              | ×       | ×       | ×      | ×      |

- ① Polyurethane rubber  
Urethane is cross linked after addition of 10~20<sub>wt</sub>% of PEL.  
⇒ Surface resistance  $10^{10}\Omega$  (Volume resistance  $10^9\Omega/\text{cm}$ )
  
- ② Urethane acrylic coating  
10<sub>wt</sub>% of PEL to the resin  
⇒ The surface resistance  $10^{10}\Omega$
  
- ③ Acrylic pressure sensitive adhesive  
5~8<sub>wt</sub>% of PEL to the resin  
⇒ The surface resistance  $10^{11}\Omega$



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